# **Integrated Chemistry and Physics East Chicago Central High School**

# **Integrated Chemistry and Physics**

# Units of Study

		🕓 1 5 Da
<u>Unit 1:</u>	Constant Velocity	ys
<u>Unit 2:</u>	Uniform Acceleration	🕓 15 Days
<u>Unit 3:</u>	Newton's Laws of Motion	🕓 15 Days
<u>Unit 4:</u>	Energy	🕓 15 Days
<u>Unit 5:</u>	Particle Theory of Matter	🕓 15 Days
<u>Unit 6:</u>	Describing Substances	🕓 15 Days
<u>Unit 7:</u>	Representing Chemical Change	🕓 15 Days
<u>Unit 8:</u>	Electricity and Magnetism	🕓 15 Days
<u>Unit 9:</u>	Waves	🕓 15 Days
<u>Unit 10:</u>	Nuclear Energy	🕓 15 Days

# Appendices

Appendix A: Curriculum Refinement Form

# **Standards Breakdown**

Green: Priority Standards	<b>Pink:</b> Support	ing Standard	s	Gray: A	dditional	Standards			
				UN	ITS				
	1 2	3	4	5	6	7	8	9	10
1.1         1.2         1.3         1.4         2.1         2.2         2.3         3.1         3.2         3.3         3.4         3.5         3.6         3.7         4.1         4.2         4.3         4.4         5.1         5.2         5.3         5.4         5.4         5.4         5.4         5.4         5.5         5.6         5.7         5.8         6.1         6.2         6.3         6.4         6.5         6.6         6.7         7.1         7.2         7.3         7.4         8.1         8.2         8.3         9.1         9.2						7	8	9	
9.3								•	
9.4 9.5									
STA         10.1           NDA         10.2									

STA NDA RDS

10.3					
10.4					
10.5					

Speed is the rate at which an object's position changes. It can be calculated by using the formula speed

= Distance/Time. An object's velocity describes its speed and direction of motion.

### Literacy Assessments:

- Speed & Velocity Teacher Document
- Speed & Velocity Student Document

#### **Priority Standards Supporting Standards** • ICP.1.4: Distinguish between the terms "speed," • **ICP.1.1:** Develop graphical, mathematical, and pictorial "velocity," "average speed," and "average velocity" and representations (such as a motion map) that describe determine the value of any of these measurements the relationship between the clock reading (time) and given either a graphical or mathematical position of an object moving at a constant velocity and representation. apply those representations to qualitatively and quantitatively describe the motion of an object. • ICP.1.2: Describe the slope of the graphical representation of position vs. clock reading (time) in terms of the velocity of the object moving in one dimension. • ICP.1.3: Distinguish between the terms "distance" and "displacement," and determine the value of either given a graphical or mathematical representation of position vs. clock reading (time).

## **Content Area Literacy Standards**

- 9-10.LST.4.1: Translate quantitative or technical information expressed in words in a text into visual form (e.g., a table or chart) and translate information expressed visually or mathematically (e.g., in an equation) into words
- 9-10.LST.1.1: Read and comprehend science and technical texts within a range of complexity appropriate for grades 9-10 independently and proficiently by the end of grade 10.

• 9-10.LST.2.2: Determine the central ideas or conclusions of a text; trace the text's explanation or depiction of a
complex process, phenomenon, or concept; provide an accurate, objective summary of the text.

<ul> <li>Enduring Understandings</li> <li>Calculate the speed using a given formula.</li> <li>Calculate the momentum of an object using a given formula.</li> </ul>	<ul> <li>Essential Questions</li> <li>How can we analyze the motion of objects?</li> <li>What is the relationship between an object's speed and the slope of a line on a speed-time graph.</li> <li>How are distance and displacement different?</li> </ul>
<ul> <li>Key Concepts</li> <li>I can distinguish between "speed," "velocity," "average speed," and "average velocity." (ICP.1.4)</li> <li>I can calculate the average speed of an object given a graph or mathematical representation. (ICP.1.4)</li> <li>I can identify when the speed and average speed are different given a graph or mathematical representation. (ICP.1.4)</li> <li>I can calculate the average velocity of an object given a graph or mathematical representation. (ICP.1.4)</li> <li>I can calculate the velocity of an object given a graph or mathematical representation. (ICP.1.4)</li> <li>I can calculate the velocity of an object given a graph or mathematical representation. (ICP.1.4)</li> <li>I can identify when the velocity and average velocity are different given a graph or mathematical representation. (ICP.1.4)</li> </ul>	<ul> <li>Related Concepts</li> <li>I can develop a graphical representation of an object's position and time when the object is moving at constant velocity. (ICP.1.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant velocity in order to explain the object's motion qualitatively. (ICP.1.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant velocity in order to explain the object's motion qualitatively. (ICP.1.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant velocity in order to explain the object's motion quantitatively. (ICP.1.1)</li> <li>I can develop a mathematical equation for an object's velocity, position or time when the object is moving at a constant velocity using v=d/t. (ICP.1.1)</li> <li>I can apply a mathematical equation for an object's motion when the object is at a constant velocity in order to explain the object's motion when the object is moving at a constant velocity using v=d/t. (ICP.1.1)</li> </ul>

		<ul> <li>motion when the to explain the obj</li> <li>I can develop a pposition at variou constant velocity.</li> <li>I can interpret picposition at variou constant velocity qualitatively. (ICF)</li> <li>I can interpret picposition at variou constant velocity qualitatively. (ICF)</li> <li>I can interpret picposition at variou constant velocity quantitatively. (ICF)</li> <li>I can calculate th moving in one dir graph. (ICP.1.2)</li> <li>I can determine t graphically when (ICP.1.3)</li> <li>I can determine t</li> </ul>	ctorial representations of an object's s times when the object is moving at to explain the object's motion 2.1.1) ctorial representations of an object's s times when the object is moving at to explain the object's motion CP.1.1) e velocity by using slope of an object mension based on the position time difference between distance. (ICP.1.3) he distance or displacement given a graph of position and time. he distance or displacement then given a position and time of an
Science and Engineering Process Standa			Vocabulary
<ul> <li>SEPS 2: Developing and Using Mode</li> <li>SEPS 4: Analyzing and Interpreting E</li> </ul>			<ul><li>Average</li><li>Constant Velocity</li></ul>
<ul> <li>SEPS 4: Analyzing and interpreting L</li> <li>SEPS 5: Using Mathematics and Cor</li> </ul>			Displacement
			• Distance
			<ul> <li>Final position/time</li> </ul>
			Frame of Reference
			<ul> <li>Initial position/time</li> <li>Instantaneous</li> </ul>
			Labels
			Meters
			Meters per Second
			Motion
			One dimension
			● Rise ● Run
			Seconds
			• Slope
			• Speed
			• Time
			<ul><li>Units</li><li>Velocity</li></ul>
	Supplement	al Resources	
Resources & Materials	Texts & Articles		Videos & Media
Speed Lab- Build 3 paper airplanes	Newsela Article-	How Fast Can	• You Tube video: Speed, Distance,
and calculate their speeds	Humans Run?		and Time (Corbettmath, 13:51)
	School R		
Textbook: Integrated Chemistry & Phys Chapter 2	ics by Glencoe	Formative Assessme Quiz on Speed Chapter 2 Test	ents

# Unit 2: Uniform Acceleration

Acceleration is the rate of change of velocity. It can be calculated using the formula acceleration = final velocity minus the original velocity divided by the time.

# Literacy Assessments:

<ul> <li>Priority Standards</li> <li>ICP.2.3: For an object thrown vertically, qualitatively describe or quantitatively determine the velocity and acceleration at various positions during its motion.</li> </ul>	<ul> <li>Supporting Standards</li> <li>ICP.2.1: Develop graphical, mathematical, and pictorial representations (such as a motion map) that describe the relationship between the clock reading (time) and velocity of an object moving at a constant acceleration and apply those representations to qualitatively and quantitatively describe the motion of an object in terms of its change in position or velocity.</li> <li>ICP.2.2: Describe the differences between average</li> </ul>
	velocity and instantaneous velocity and be able to determine either quantity given a graph of position vs clock reading (time).
Content Area Lit • 9-10.LST.4.1: Translate quantitative or technical informativable or chart) and translate information expressed visual • 9-10.LST.1.1: Read and comprehend science and technic grades 9-10 independently and proficiently by the end of • 9-10.LST.2.2: Determine the central ideas or conclusions complex process, phenomenon, or concept; provide an a	lly or mathematically (e.g., in an equation) into words cal texts within a range of complexity appropriate for grade 10. of a text; trace the text's explanation or depiction of a
Enduring Understandings	Essential Questions
Calculate the acceleration using a given formula.	<ul> <li>How are acceleration, time, and velocity related?</li> <li>How can an object's acceleration be calculated?</li> </ul>
<ul> <li>Key Concepts</li> <li>I can describe the velocity qualitatively at various positions of an object that is launched vertically. (ICP.2.3)</li> <li>I can describe the velocity quantitatively at various positions of an object that is launched vertically using a = v/t and a=g=9.8 m/s^2. (ICP.2.3)</li> <li>I can explain the acceleration both qualitatively (constant) and quantitatively (-9.8 m/s^2) at various positions of an object that is launched vertically. (ICP.2.3)</li> <li>(ICP.2.3)</li> </ul>	<ul> <li>Related Concepts</li> <li>I can develop a graphical representation of an object's position and time when the object is moving at constant acceleration. (ICP.2.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant acceleration in order to explain the object's motion qualitatively. (ICP.2.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant acceleration in order to explain the object's motion qualitatively. (ICP.2.1)</li> <li>I can interpret a graphical representation of an object's position and time when the object is moving at a constant acceleration in order to explain the object's motion quantitatively. (ICP.2.1)</li> <li>I can develop a mathematical equation for an object's acceleration, velocity, or time when the object is moving at a constant acceleration. (ICP.2.1)</li> <li>I can apply a mathematical equation for an object's motion when the object is at a constant acceleration in order to explain the object's motion qualitatively. (ICP.2.1)</li> <li>I can apply a mathematical equation for an object's motion when the object is at a constant acceleration in order to explain the object's motion qualitatively. (ICP.2.1)</li> <li>I can apply a mathematical equation for an object's motion when the object is at a constant acceleration in order to explain the object's motion qualitatively. (ICP.2.1)</li> <li>I can develop a pictorial representation of an object's position at various times when the object is moving at constant acceleration. (ICP.2.1)</li> <li>I can interpret pictorial representations of an object's position at various times when the object is moving at constant acceleration. (ICP.2.1)</li> </ul>

		<ul> <li>qualitatively. (ICF</li> <li>I can interpret pice position at various constant accelerate quantitatively. (ICF)</li> <li>I can calculate the traveling under us (ICP.2.2)</li> <li>I can calculate the traveling under us (ICP.2.2)</li> <li>I can identify when average velocity</li> </ul>	ctorial representations of an object's is times when the object is moving at ation to explain the object's motion CP.2.1) e average velocity of an object niform acceleration given a graph. e instantaneous velocity of an object niform acceleration given a graph. en the instantaneous velocity and are different for an object traveling celeration given a graph. (ICP.2.2)
<ul> <li>SEPS 2: Developing and Using Mode</li> <li>SEPS 4: Analyzing and Interpreting E</li> <li>SEPS 5: Using Mathematics and Con</li> </ul>	Data nputational Thinking		Vocabulary • Acceleration • Average Velocity • Gravity (concept and value) • Horizontal • Instantaneous Velocity • Meters per second squared (m/s^2) • Vertical
	Supplement	al Resources	
<ul> <li>Resources &amp; Materials</li> <li>Mini-Lab -Determine the Direction of Acceleration (pg 57 in textbook)</li> </ul>	Texts & Articles • Newsela Article- Historian Reflects Change		<ul> <li>Videos &amp; Media</li> <li>You Tube Video: What is Acceleration? (2.01min)</li> </ul>
	School R	esources	
Textbook Chapter 2		Formative Assessm Quiz on Acceleratio Chapter 2 Test	

Unbalanced forces change motion. Discuss Newton's 3 laws of motion to relate the changes in an object's motion with the forces acting on it. Newton's Laws can be used to explain everyday events, such as falling and collisions.

<ul> <li>Priority Standards</li> <li>ICP.3.1: Develop pictorial and graphical representations which show that a single external applied force changes the velocity of an object, and that when no force acts, the velocity of an object remains constant.</li> <li>ICP.3.4: Develop pictorial and graphical</li> </ul>	<ul> <li>Supporting Standards</li> <li>ICP.3.2: Construct force diagrams and combine forces to determine the equivalent single net force acting on the object when more than one force is acting on the object.</li> <li>ICP.3.3: Distinguish between forces acting on a body and forces exerted by the body. Categorize forces as</li> </ul>
<ul> <li>representations which show that a non-zero net force on an object results in an acceleration of the object and that the acceleration of an object of constant mass is proportional to the total force acting on it, and inversely proportional to its mass for a constant applied total force.</li> <li>ICP.3.7: Develop pictorial and graphical representations which show that when two objects interact, the forces occur in pairs according to Newton's third law and that the change in motion of each object is dependent on the mass of each object.</li> </ul>	<ul> <li>ICP.3.5: Qualitatively describe and quantitatively determine the magnitude and direction of forces from observing the motion of an object of known mass.</li> <li>ICP.3.6: Qualitatively describe and quantitatively determine the acceleration of an object of known mass from observing the forces acting on that object.</li> </ul>
Content Area Lit	ceracy Standards
<ul> <li>9-10.LST.3.2: Analyze the structure of the relationships at key terms</li> <li>9-10.LST.4.1: Translate quantitative or technical informati table or chart) and translate information expressed visual</li> <li>9-10.LST.2.2: Determine the central ideas or conclusions complex process, phenomenon, or concept; provide an a</li> </ul>	on expressed in words in a text into visual form (e.g., a ly or mathematically (e.g., in an equation) into words of a text; trace the text's explanation or depiction of a
Enduring Understandings	Essential Questions
<ul> <li>Newton's First Law of Motion states that an object moves at a constant velocity unless an unbalanced force acts on it. (Law of Inertia)</li> <li>Newton's Second Law of Motion states an object's acceleration is in the same direction as the net force on the object and is equal to the net force exerted on it, divided by its mass. (a = F/m)</li> <li>Newton's Third Law of Motion states that when one object exerts a force on the first that is equal in strength and opposite in direction.</li> </ul>	<ul> <li>How can one explain and predict interactions between objects and within systems of objects?</li> <li>How can one predict an object's continued motion, change in motion, or stability?</li> </ul>
Key Concepts	Related Concepts
<ul> <li>I can draw a pictorial representation demonstrating the velocity of an object experiencing a single uniform force. (ICP.3.1)</li> <li>I can develop a graph of an object's velocity when the object is experiencing a single uniform force. (ICP.3.1)</li> <li>I can develop a graph that demonstrates the acceleration of an object as it relates to the force applied to the object with a constant mass. (ICP.3.4)</li> <li>I can develop a pictorial representation that demonstrates the acceleration of an object with a constant mass. (ICP.3.4)</li> </ul>	<ul> <li>I can develop a force diagram illustrating how multiple forces acting on an object simplify into one net force. (ICP.3.2)</li> <li>I can calculate the value of one equivalent net force for an object that experiences multiple forces. (ICP.3.2)</li> <li>I can distinguish between forces acting on a body and forces exerted by the body. (ICP.3.3)</li> <li>I can categorize forces as contact, friction or field forces based on a given description or pictorial information. (ICP.3.3)</li> <li>I can qualitatively describe the magnitude and direction of forces from observing the motion of an object of known mass. (ICP.3.5)</li> </ul>

<ul> <li>I can develop a graph that demonstrative acceleration of an object is proportion force acting on it. (ICP.3.4)</li> <li>I can develop a graph that demonstrative acceleration of an object as it inverses mass of an object. (ICP.3.4)</li> <li>I can develop pictorial and graphical which show that when two objects into occur in pairs according to Newton's</li> <li>I can develop pictorial and graphical which show that the change in motion dependent on the mass of each object</li> </ul>	hal to the total ates the ely related to the representations teract, the forces third law. (ICP.3.7) representations n of each object is	direction of force object of known r I can qualitatively of known mass fr object. (ICP.3.6) I can quantitative	y describe the acceleration of an object om observing the forces acting on that ly determine the acceleration of an mass from observing the forces acting
Science and Engineering Process Standa • A			Vocabulary Acceleration Action Applied force Constant Contact forces Direction External Field forces Force Force Diagrams Friction Inversely Proportional Kilogram Magnitude Mass Motion Net Force Newton's First and Second Laws Newton's First Law Newton's Second Law Newton's Third Law Observing Motion Reaction Unit Newton Velocity
	Supplementa	al Resources	
Resources & Materials	Texts & Articles		Videos & Media
<ul> <li>Lab on Newton's First Law-Law of Inertia</li> </ul>	<ul> <li>Newsela Article Newton's Laws of</li> </ul>	f Motion?	<ul> <li>Understanding Car Crashes: it's Basic Physics</li> </ul>
	School R		
Textbook Chapter 3		Formative Assessme Quiz on Newton's 2 Quiz on all of Newt Chapter 3 Test	nd Law

### Unit 4: Energy

### **General Description of the Unit**

Work is force applied through a distance and machines make doing work easier or faster by changing the force needed to do the work. Energy is the ability to cause change. Kinetic energy is the energy of motion, while potential energy is stored energy due to the interactions between objects. Forms of energy include mechanical, electrical, chemical, thermal, and radiant energy. Energy cannot be created or destroyed. Temperature is a measure of the average kinetic energy of the particles of an object. There are 3 ways to transfer heat energy: conduction, convection, and radiation.

## Literacy Assessments:

- <u>Heat Transfer Teacher Document</u>
- <u>Heat Transfer Student Document</u>

#### **Priority Standards Supporting Standards** • ICP.4.1: Define energy as a quantity that can be • ICP.4.4: Qualitatively and quantitatively analyze various scenarios to describe how energy may be transferred represented as being within a system that is distinct from the remainder of the universe and is measured in into or out of a system by doing work through an external force or adding or removing heat. Joules. **ICP.5.5:** Evaluate graphical or pictorial • **ICP.4.2**: Identify forms of energy present in a system representations that describe the relationship among (kinetic, gravitational, elastic, etc.), and pictorially the volume, temperature, and number of molecules represent the distribution of energies, such as using pie and the pressure exerted by the system to or bar charts. qualitatively and quantitatively describe how • ICP.4.3: Understand and explain that the total energy in changing any of those variables affects the others. a closed system is conserved. **ICP.5.4:** Distinguish "temperature" from "thermal energy," compare and contrast the Fahrenheit, Celsius, and Kelvin temperature scales, and convert temperatures between them. ICP.5.3: At the particle level, describe the relationship between temperature and the average kinetic energy of particles in the system and describe how a thermometer measures the temperature of a system.

### **Content Area Literacy Standards**

- 9-10.LST.4.1: Translate quantitative or technical information expressed in words in a text into visual form (e.g., a table or chart) and translate information expressed visually or mathematically (e.g., in an equation) into words
- 9-10.LST.1.1: Read and comprehend science and technical texts within a range of complexity appropriate for grades 9-10 independently and proficiently by the end of grade 10.
- 9-10.LST.2.2: Determine the central ideas or conclusions of a text; trace the text's explanation or depiction of a complex process, phenomenon, or concept; provide an accurate, objective summary of the text.

<ul> <li>Enduring Understandings</li> <li>Work is force applied through a distance and is measured in Joules.</li> <li>Kinetic energy is energy of motion.</li> <li>Potential energy is energy that is stored due to the interactions between objects.</li> <li>The law of conservation of energy states that energy cannot be created or destroyed, it can be converted from one form to another or transferred from one place to another.</li> </ul>	<ul> <li>Essential Questions</li> <li>How does the flow of energy affect the materials in the system?</li> <li>How is energy transferred and conserved?</li> <li>What is the ability of energy?</li> </ul>
<ul> <li>Key Concepts</li> <li>I can qualitatively and quantitatively analyze various scenarios to describe how energy may be transferred into or out of a system by doing work through an external force or adding or removing heat. (ICP.4.4)</li> </ul>	<ul> <li>Related Concepts</li> <li>I can define energy as a quantity that can be represented as being within a system that is distinct from the remainder of the universe and is measured in Joules. (ICP.4.1)</li> </ul>

• I can quantitatively analyze various scenarios to describe how energy may be transferred into or out of a system by doing work through an external force or adding or removing heat. (ICP.4.4)		<ul> <li>(kinetic, gravitation)</li> <li>I can pictorially resurct as using piese</li> <li>I can explain that conserved. (ICP.4)</li> <li>I can described temperature of</li> <li>I can distinguist (ICP.5.4)</li> <li>I can compare and Kelvin temperature</li> </ul>	how a thermometer measures the a system. (ICP.5.3) h temperature from thermal energy. and contrast the Fahrenheit, Celsius, perature scales. (ICP.5.4) emperatures between the various	
Science and Engineering Process Standards • SEPS 2: Developing and Using Models • SEPS 5: Using Mathematics and Computational Thinking • SEPS 7: Engaging in Argument from Evidence			Vocabulary • Conservation of Energy • Elastic Potential Energy • Energy • External Force • Gravitational Potential Energy • Heat • Joules • Kinetic Energy • System	
Supplemental Resources				
Resources & Materials • Popcorn Lab- 3 Ways to Cook Popcorn to Demonstrate the 3 Types of Heat Transfer • Lab on Work and Power	Texts & Articles • Newsela Articles • Six Kinds of Simple Machines • Matter and Energy- What is Energy? • What is Heat Energy? School Resources		Videos & Media • You Tube video: WCLN-Physics-Energy 4-Calculating Work	
	School R			
Textbook Chapter 4 Chapter 5		Formative Assessme Quiz on Work Quiz on Heat Trans Test on Chapter 4 Test on Chapter 5		

Matter can exist as a solid, liquid, gas, or plasma. The atoms and molecules that make up matter are in continuous, random motion.

<ul> <li>Priority Standards</li> <li>ICP.5.1: Develop pictorial representations which show that matter is made of particles.</li> </ul>	<ul> <li>Supporting Standards</li> <li>ICP.5.2: Describe the assumptions used to develop the kinetic theory of gasses.</li> <li>ICP.5.6: Describe and demonstrate how the kinetic theory can be extended to describe the properties of liquids and solids by introducing attractive forces between the particles.</li> <li>ICP.5.7: Analyze a heating / cooling curve to describe how adding or removing thermal energy from a system changes the temperature or state of an object and be able to identify the melting and freezing temperatures of the system.</li> <li>ICP.5.8: Collect and use experimental data to determine the number of items in a sample without actually counting them and qualitatively relate this to Avogadro's hypothesis.</li> </ul>
Content Area Li	teracy Standards
<ul> <li>9-10.LST.4.1: Translate quantitative or technical informativative or chart) and translate information expressed visual</li> <li>9-10.LST.1.1: Read and comprehend science and technic grades 9-10 independently and proficiently by the end of</li> <li>9-10.LST.2.2: Determine the central ideas or conclusions complex process, phenomenon, or concept; provide an and provide provide provide and provide prov</li></ul>	ion expressed in words in a text into visual form (e.g., a lly or mathematically (e.g., in an equation) into words cal texts within a range of complexity appropriate for grade 10. of a text; trace the text's explanation or depiction of a
<ul> <li>Enduring Understandings</li> <li>Matter can exist as a solid, liquid, gas. or plasma.</li> <li>The kinetic theory is an explanation of how the particles in gases behave.</li> <li>Gases respond to changes in pressure, temperature, and volume in predictable ways.</li> </ul>	<ul> <li>Essential Questions</li> <li>Which state of matter do you believe is the most important for survival?</li> <li>How does temperature affect matter?</li> </ul>
Key Concepts	Related Concepts
<ul> <li>I can create a pictorial representation which shows that matter is made of particles. (ICP.5.1)</li> <li>I can evaluate graphical or pictorial representations that describe the relationship among the volume, temperature, and number of molecules and the pressure exerted by the system to qualitatively describe how changing any of those variables affects the others. (ICP.5.5)</li> <li>I can evaluate graphical or pictorial representations that describe the relationship among the volume, temperature, and number of molecules and the pressure exerted by the system to quantitatively describe the relationship among the volume, temperature, and number of molecules and the pressure exerted by the system to quantitatively describe how changing any of those variables affects the others. (ICP.5.5)</li> </ul>	<ul> <li>I can describe the assumptions used to develop the kinetic theory of gasses. (ICP.5.2)</li> <li>I can describe at the particle level the relationship between temperature and the average kinetic energy of particles in the system. (ICP.5.3)</li> <li>I can describe how the kinetic theory can be extended to describe the properties of liquids and solids by introducing attractive forces between the particles. (ICP.5.6)</li> <li>I can analyze a heating / cooling curve to describe how adding or removing thermal energy from a system changes the temperature or state of an object. (ICP.5.7)</li> <li>I can identify the melting and freezing temperatures of the system. (ICP.5.7)</li> <li>I can collect experimental data to determine the number of items in a sample without actually counting them. (ICP.5.8)</li> <li>I can relate collected experimental data qualitatively to Avogadro's hypothesis. (ICP.5.8)</li> </ul>

Science and Engineering Process Standa	ards	Vocabulary
Science and Engineering Process Standa • SEPS 3: Planning and Carrying Out • SEPS 4: Analyzing and Interpreting I	Investigations	Vocabulary Average Kinetic Energy Avogadro's Hypothesis Celsius Fahrenheit Freezing Temperature Heating / Cooling Curve Kelvin Kinetic Kinetic Theory Kinetic Theory of Gasses Liquids Matter Melting Temperature Molecules Particles Pictorial Representation Pressure Sample Solids State System System Changes
		<ul> <li>Temperature</li> <li>Thermal Energy</li> <li>Thermometer</li> <li>Volume</li> </ul>
	Supplemental Resources	
Resources & Materials • Lab- Phase Changes	Texts & Articles <ul> <li>Newsela Article- The 3 Phases of Matter: Solids. Liquids, and Gases</li> </ul>	<ul> <li>Videos &amp; Media</li> <li>You Tube Video: Phase Changes_ Middle School Science (7:40 min)</li> <li>You Tube Video: Phases of Matter and Phase Changes (6:06 min)</li> </ul>
	School Resources	
Textbook Chapter 14	Formative Assessm Quiz over Solids, L Chapter 14 Test	

Matter exists as either a pure substance or a mixture. A physical property can be observed without changing the identity of the material; a chemical property can be observed when one or more new substances are formed. The properties of an element are determined by the structure of its atoms.

# Literacy Assessments:

- <u>Molecules, Mixtures, & Compounds Teacher Document</u>
  <u>Molecules, Mixtures, & Compounds Student Document</u>

Priority Standards	Supporting Standards
<ul> <li>ICP.6.1: Distinguish between elements, mixtures, and compounds based on their composition and bonds and be able to construct or sketch particle models to represent them.</li> <li>ICP.6.4: Given the periodic table, determine the atomic mass, atomic number, and charges for any element.</li> </ul>	<ul> <li>ICP.6.2: Develop graphical and mathematical representations to show that mixtures can be made in any proportion and separated based on the properties of the components of the mixture and apply those representations to quantitatively determine the ratio of components.</li> <li>ICP.6.3: Cite the evidence that supports the idea that some pure substances are combined of elements in a definite ratio, as for example seen in electrolysis of water.</li> <li>ICP.6.5: Given a periodic table, understand and describe the significance of column location for the elements by calculation of molar ratios of known compounds.</li> <li>ICP.6.6: Develop graphical and mathematical representations that describe the relationship between volume and mass of an object, describe the slope in terms of the object's density, and apply those representations to qualitatively and quantitatively determine the mass or volume of any object.</li> <li>ICP.6.7: Describe how both density and molecular structure are applicable in distinguishing the properties</li> </ul>
	• ICP.6.7: Describe how both density and molecular
	of gases from those of liquids and solids.
	teracy Standards
<ul> <li>9-10.LST.1.1: Read and comprehend science and techni grades 9-10 independently and proficiently by the end of</li> <li>9-10.LST.2.2: Determine the central ideas or conclusions complex process, phenomenon, or concept; provide an a</li> </ul>	grade 10. of a text; trace the text's explanation or depiction of a
Enduring Understandings	Essential Questions
<ul> <li>Matter exists as either a pure substance or a mixture.</li> <li>Protons and neutrons form the nucleus of an atom, and</li> </ul>	<ul> <li>How can one explain the structure, properties and interactions of matter?</li> </ul>
<ul> <li>electrons occupy a space surrounding the nucleus.</li> <li>All atoms of the same element have the same number of protons but can have different numbers of neutrons.</li> <li>Atoms of elements that are in the same group of the periodic table have similar physical and chemical properties.</li> </ul>	How do particles combine to form the variety of matter one observes?
Key Concepts	Related Concepts
<ul> <li>I can distinguish between elements, mixtures, and compounds based on their composition and bonds. (ICP.6.1)</li> <li>I can construct or sketch particle models to represent</li> </ul>	• I can develop graphical representations to show that mixtures can be made in any proportion and separated based on the properties of the components of the mixture. (ICP.6.2)
them. (ICP.6.1)	• I can develop mathematical representations to show that mixtures can be made in any proportion and

<ul> <li>I can determine the atomic mass, ator charges for any element given a period (ICP.6.4)</li> <li>Science and Engineering Process Standa</li> <li>SEPS 6: Constructing Explanations a</li> <li>SEPS 7: Engaging in Argument from</li> </ul>	odic table. of the mixture. I can apply tho determine the if I can cite the e pure substance ratio, as for exa (ICP.6.3) I can describe the elements b compounds giv I can develop giv the relationship (ICP.6.6) I can develop r describe the re an object (ICP. I can describe (ICP.6.6) I can describe the relationship (ICP.6.6) I can describe the relationship (ICP.6.6) I can describe the re an object (ICP. I can describe the relationship (ICP.6.6) I can describe the relationship (ICP.6.6) I can describe (ICP.6.6) I can apply tho quantitatively co object. (ICP.6.6) I can describe structure are a of gases from the rds and Designing Solutions	se representations to quantitatively ratio of components. (ICP.6.2) vidence that supports the idea that some es are combined of elements in a definite ample seen in electrolysis of water. The significance of column location for y calculation of molar ratios of known ren a periodic table. (ICP.6.5) graphical representations that describes between volume and mass of an object nathematical representations that lationship between volume and mass of 6.6) the slope in terms of the object's density. se representations to qualitatively and etermine the mass or volume of any
	Supplemental Resources	
Resources & Materials <ul> <li>The Periodic Table Lab</li> </ul>	Texts & Articles <ul> <li>Newsela Article- The Periodic</li> </ul>	Videos & Media • You Tube video: The Periodic Table
	Table: A Classic Design	• You Tube Video: The Periodic Table Explained: Introduction
	School Resources	
Textbook Chapter 15 Chapter 16 Chapter 17	Chapter 15 Test Quiz over Struct	ments sition of matter-Chapter 15. ıre of Atom - Chapter 16 gram Quiz- Chapter 16

Quiz over Metals and Nonmetals Chapter 17 Test

A chemical reaction involves changing one or more substances into a different substance or substances. A balanced chemical equation describes the rearrangement of atoms in a chemical reaction. Reactions can be classified into 5 main categories based on how atoms are rearranged.

### Literacy Assessments:

- <u>Conservation of Energy Teacher Document</u>
- Conservation of Energy Student Document
- <u>Chemical Reactions Teacher Document</u>
- Chemical Reactions Student Document

Priority Standards	Supporting Standards
• ICP.7.1: Pictorially or mathematically represent chemical changes using particle diagrams and chemical equations.	<ul> <li>ICP.7.2: Demonstrate the Law of Conservation of Matter in terms of atoms and mass of substances by balancing equations.</li> <li>ICP.7.3: Differentiate the basic types of reactions, for example: synthesis, decomposition, combustion, single replacement, and double replacement.</li> <li>ICP.7.4: Using balanced equations and stoichiometric calculations, demonstrate the principle of Conservation of Matter in terms of atoms and mass.</li> </ul>

### **Content Area Literacy Standards**

- 9-10.LST.1.1: Read and comprehend science and technical texts within a range of complexity appropriate for grades 9-10 independently and proficiently by the end of grade 10.
- 9-10.LST.2.2: Determine the central ideas or conclusions of a text; trace the text's explanation or depiction of a complex process, phenomenon, or concept; provide an accurate, objective summary of the text.
- 9-10.LST.3.1: Determine the meaning of symbols, key terms, and other domain-specific words and phrases as they are used in a specific scientific or technical context relevant to grades 9-10 texts and topics.

<ul> <li>Enduring Understandings</li> <li>Atoms form ionic bonds by transferring electrons and form covalent bonds by sharing electrons.</li> <li>For both ionic and covalent compounds, you can write a name from the chemical formula ro a chemical formula from the name.</li> <li>There are 5 main categories of chemical reactions: combustion, synthesis, decomposition, single displacement, and double displacement.</li> </ul>
<ul> <li>form covalent bonds by sharing electrons.</li> <li>For both ionic and covalent compounds, you can write a name from the chemical formula ro a chemical formula from the name.</li> <li>There are 5 main categories of chemical reactions: combustion, synthesis, decomposition, single</li> </ul>
<ul> <li>displacement, and double displacement.</li> <li>A balanced chemical equation describes the rearrangement of atoms in a chemical reaction.</li> </ul>
<ul> <li>Key Concepts</li> <li>I can pictorially or mathematically represent chemical changes using particle diagrams and chemical equations. (ICP.7.1)</li> <li>I can demonstrate the Law of Conservation of Matterms of atoms and mass of substances by balance equations. (ICP.7.2)</li> <li>I can tell the difference between the basic types of reactions: synthesis, decomposition, combustion, single replacement, and double replacement. (ICP)</li> <li>I can demonstrate the principle of Conservation of Matterms of atoms and mass of substances by balance equations. (ICP.7.2)</li> </ul>
Matter in terms of atoms and mass using balanced
Matter in terms of atoms and mass using balanced equations and stoichiometric calculations. (ICP.7.4

	Sumplemente	1 Datonwood	<ul> <li>Balancing Equations</li> <li>Chemical Changes</li> <li>Chemical Equations</li> <li>Conservation of Matter</li> <li>Decomposition</li> <li>Double Replacement</li> <li>Law of Conservation of Matter</li> <li>Mass</li> <li>Single Replacement</li> <li>Stoichiometric Calculations</li> <li>Synthesis</li> </ul>
December 9 Material	Supplementa	Intestunces	Videos & Media
<ul> <li>Resources &amp; Materials</li> <li>Lab: Strength of Attraction:lons vs Molecules pg 572 in textbook</li> </ul>	Texts & Articles • Newsela Article: lo Bonding: Understa Difference		<ul> <li>You Tube video: Ionic vs. Covalent Bonds</li> </ul>
Textbook Chapter 18 Chapter 19		Formative Assessm Quiz over ionic and Chapter 18 Test Quiz over types of Chapter 19	l covalent bonding

Electricity consists of static and moving electric charges. Our ability to manipulate electric charge has allowed us to produce everything from electric lights to digital music players to ultraportable computers.

## Literacy Assessments:

- <u>Electrical Current Teacher Document</u>
- <u>Electrical Current Student Document</u>

<ul> <li>Supporting Standards</li> <li>ICP.8.2: Describe the relationship among voltage, current, and resistance for an electrical system consisting of a single voltage source and a single device.</li> <li>ICP.8.3: Describe on a macroscopic scale how any distribution of magnetic materials (e.g. iron filings, ferrofluid, etc.) aligns with the magnetic field created by a simple magnet.</li> </ul>
teracy Standards
cal texts within a range of complexity appropriate for grade 10. of a text; trace the text's explanation or depiction of a accurate, objective summary of the text. rms, and other domain-specific words and phrases as they ant to grades 9-10 texts and topics.
<ul> <li>Essential Questions</li> <li>How can objects become electrically charged?</li> <li>How does Ohm's law relate current, voltage difference, and resistance?</li> </ul>
<ul> <li>Related Concepts</li> <li>I can describe the relationship among voltage, current, and resistance for an electrical system consisting of a single voltage source and a single device. (ICP.8.2)</li> <li>I can describe on a macroscopic scale how any distribution of magnetic materials (e.g. iron filings, ferrofluid, etc.) aligns with the magnetic field created by a simple magnet. (ICP.8.3)</li> </ul>
ions rmation  Vocabulary  Current  Electrical Current  Electrical System  Electrons  Ferrofluid  Macroscopic Scale  Magnet  Magnetic Field  Magnetic Materials  Resistance  Voltage  Voltage Voltage Device  Voltage Source

Resources & Materials • Lab on Conductors & Insulators	Electricity Provid Used to Power M World	ies; What is Static Relationship ity and Magnetism es the Energy lost of the Modern	<ul> <li>Videos &amp; Media</li> <li>You Tube video: Introduction to Electricity-Video for Kids (5:25)</li> </ul>
	School F	Resources	
Chapter 6		Formative Assessm Quiz over Electricit Chapter 6 Test	

Waves are one of many ways to transfer energy. All types of waves, including water waves, waves on a rope, or spring, sound waves, and light waves, transfer energy without transferring matter.

<ul> <li>Priority Standards</li> <li>ICP.9.1: Develop qualitative particle models of mechanical waves and explain the relationship of the particles and their interactions in transverse and longitudinal waves, as well as, how waves appear in nature as in water waves and tsunamis, ground waves in earthquakes, and sound waves.</li> </ul>	<ul> <li>Supporting Standards</li> <li>ICP.9.2: Develop and apply a simple mathematical model regarding the relationship among frequency, wavelength, and speed of waves in a medium as well.</li> <li>ICP.9.3: Qualitatively describe the reflection and transmission of a mechanical wave at either a fixed or free boundary or interface.</li> <li>ICP.9.4: Describe how interacting waves produce different phenomena than singular waves in a medium (e.g. periodic changes in volume of sound or resonance).</li> <li>ICP.9.5: Describe and provide examples of how modern technologies use mechanical or electromagnetic waves and their interactions to transmit information.</li> </ul>
Content Area Lit	
<ul> <li>9-10.LST.1.1: Read and comprehend science and technic grades 9-10 independently and proficiently by the end of</li> <li>9-10.LST.2.2: Determine the central ideas or conclusions complex process, phenomenon, or concept; provide an a</li> <li>9-10.LST.3.1: Determine the meaning of symbols, key ter are used in a specific scientific or technical context relevant</li> </ul>	cal texts within a range of complexity appropriate for grade 10. of a text; trace the text's explanation or depiction of a ccurate, objective summary of the text. ms, and other domain-specific words and phrases as they
Enduring Understandings	Essential Questions
<ul> <li>Wave properties depend on the vibrations of the wave source and the material in which the wave travels.</li> <li>Calculate the wave speed using a given formula.</li> </ul>	<ul> <li>How are waves used to transfer energy and information?</li> <li>What are the various methods for creating waves?</li> </ul>
<ul> <li>Key Concepts</li> <li>I can construct qualitative particle models of mechanical waves. (ICP.9.1)</li> <li>I can explain the relationship of the particles and their interactions in transverse and longitudinal waves. (ICP.9.1)</li> <li>I can explain how waves appear in nature as in water waves and tsunamis, ground waves in earthquakes, and sound waves. (ICP.9.1)</li> <li>Science and Engineering Process Standards</li> </ul>	<ul> <li>Related Concepts</li> <li>I can develop a simple mathematical model regarding the relationship among frequency, wavelength, and speed of waves in a medium. (ICP.9.2)</li> <li>I can apply a simple mathematical model regarding the relationship among frequency, wavelength, and speed of waves in a medium. (ICP.9.2)</li> <li>I can qualitatively describe the reflection and transmission of a mechanical wave at a fixed boundary. (ICP.9.3)</li> <li>I can qualitatively describe the reflection and transmission of a mechanical wave at a free boundary. (ICP.9.3)</li> <li>I can qualitatively describe the reflection and transmission of a mechanical wave at a free boundary. (ICP.9.3)</li> <li>I can qualitatively describe the reflection and transmission of a mechanical wave at an interface. (ICP.9.3)</li> <li>I can describe how interacting waves produce different phenomena than singular waves in a medium. (ICP.9.4)</li> <li>I can describe how modern technologies use mechanical or electromagnetic waves and their interactions to transmit information. (ICP.9.5)</li> <li>I can provide examples of how modern technologies use mechanical or electromagnetic waves and their interactions to transmit information. (ICP.9.5)</li> </ul>

<ul> <li>SEPS 2: Developing and Using Mode</li> <li>SEPS 5: Using Mathematics and Cor</li> <li>SEPS 6: Constructing Explanations a</li> <li>SEPS 8: Obtaining, Evaluating, and 0</li> </ul>	<ul> <li>Fixed Boundary</li> <li>Free Boundary</li> <li>Frequency</li> <li>Interacting Waves</li> <li>Interface</li> <li>Longitudinal Waves</li> <li>Mechanical Wave</li> <li>Medium</li> <li>Particles</li> <li>Reflection</li> <li>Resonance</li> <li>Transmission</li> <li>Transverse Waves</li> <li>Wavelength</li> <li>Mechanical Waves</li> <li>Electromagnetic Waves</li> <li>Transmit</li> </ul>				
Supplemental Resources					
Resources & Materials • Lab- Wave Speed and Tension Pg 285 in Textbook Textbook Chapter 9	Texts & Articles • Newsela Articles: • Wave Properties • How Do Waves Behave? • Measurable Properties of Wa School Resources Formative A Quiz over v Chapter 9 Te	<b>Assessments</b> vaves			

The energy resources that we use can be renewable or nonrenewable. Our use of these resources impacts our daily lives and the environment in which we live.

## Literacy Assessments:

- <u>Atomic Models Teacher Document</u>
- <u>Atomic Models Student Document</u>
- <u>The Atomic Nucleus Teacher Document</u>
- <u>The Atomic Nucleus Student Document</u>

#### **Priority Standards Supporting Standards** • ICP.10.2: Describe the model of the atomic nucleus • ICP.10.1: Describe and compare/contrast the atomic models suggested by Rutherford and Bohr. and explain how the nucleus stays together in spite of the repulsion between protons. • **ICP.10.3**: Develop and apply simple qualitative models or sketches of the atomic nucleus that illustrate nuclear structures before and after undergoing fusion, fission, or radioactive decay. • ICP.10.4: Distinguish between fusion, fission, and radioactivity and gualitatively compare the amount of energy released in these processes. • ICP.10.5: Explain the potential applications and possible consequences as the result of nuclear processes such as the generation of energy at nuclear power plants, including the potential damage that radioactivity can cause to biological tissues. **Content Area Literacy Standards** • 9-10.LST.1.1: Read and comprehend science and technical texts within a range of complexity appropriate for grades 9-10 independently and proficiently by the end of grade 10. • 9-10.LST.2.2: Determine the central ideas or conclusions of a text; trace the text's explanation or depiction of a complex process, phenomenon, or concept; provide an accurate, objective summary of the text. **Enduring Understandings Essential Ouestions** Nuclear power plants transform nuclear energy into What forces hold nuclei together and mediate nuclear electrical energy. processes? • Renewable energy resources help lesson human What distinguishes a nuclear reaction from an ordinary dependence on fossil fuels. chemical reaction? **Key Concepts Related Concepts** • I can describe the model of the atomic nucleus. • I can describe the atomic models suggested by Rutherford and Bohr. (ICP.10.1) (ICP.10.2) • I can explain how the nucleus stays together in spite of • I can compare/contrast the atomic models suggested the repulsion between protons. (ICP.10.2) by Rutherford and Bohr. (ICP.10.1) • I can develop simple qualitative models or sketches of the atomic nucleus that illustrate nuclear structures before and after undergoing fusion, fission, or radioactive decay. (ICP.10.3) • I can apply simple qualitative models or sketches of the atomic nucleus that illustrate nuclear structures before and after undergoing fusion, fission, or radioactive decay. (ICP.10.3) • I can distinguish between fusion, fission, and radioactivity. (ICP.10.4) • I can qualitatively compare the amount of energy released in these processes. (ICP.10.4) • I can explain the potential applications and possible consequences as the result of nuclear processes such

Science and Engineering Process Standa • SEPS 1: Asking Questions and Defin • SEPS 2: Developing and Using Mode • SEPS 8: Obtaining, Evaluating, and e	iing Problems els	including the pote cause to biologic	n of energy at nuclear power plants, ential damage that radioactivity can al tissues. (ICP.10.5) Vocabulary • Atomic Models • Atomic Nucleus • Biological Tissues • Bohr • Consequences • Energy • Fission • Fusion • Nuclear Power Plants • Nuclear Processes • Protons • Radioactive Decay • Radioactivity • Rutherford	
Supplemental Resources				
Resources & Materials •	<ul> <li>Texts &amp; Articles</li> <li>Newsela Articles:</li> <li>The Sun, an Engine of Nuclear Energy</li> <li>Nuclear Power as an Energy Source Has its Pros and Cons</li> <li>How Atomic and Thermonuclear Bombs work</li> </ul>		Videos & Media • :YouTube video: Nuclear Energy Explained:How Does it Work? (Kurzgesagt, 5:18)	
School Resources				
Textbook Chapter 8		Formative Assessm Quiz over chapter & Chapter 8 Test		